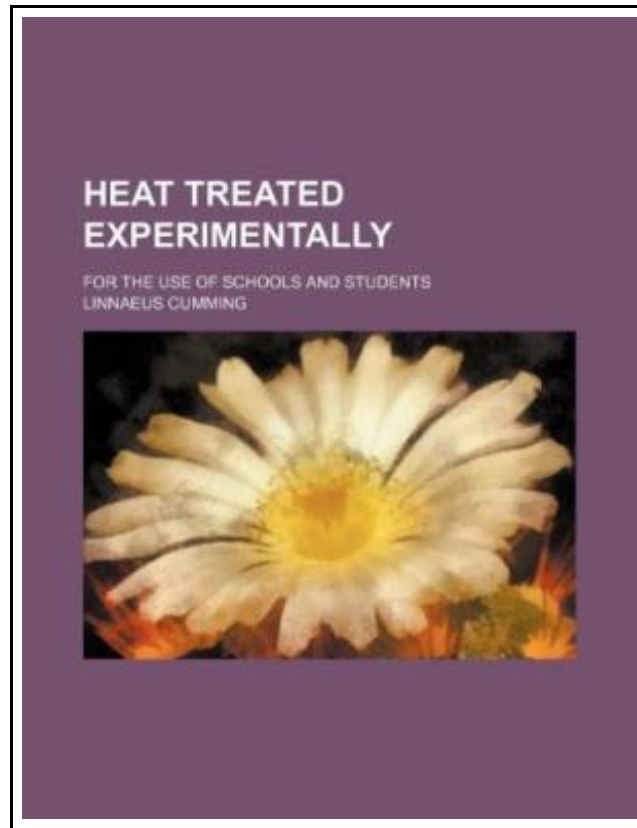


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Rarebookclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1894 Excerpt: .of heat, and have said nothing about the temperature. It may be thought desirable to give some proof that the mechanical equivalent of a certain quantity of heat is the same, whatever be the temperature of the substance in which the heat is contained. Such an experiment can easily be arranged by means of an ice calorimeter. If a weighed quantity of water or mercury at 80 C. be placed in an ice calorimeter, and the quantity of ice melted be observed, and if further eight times that quantity of water be taken at 10 C. and be placed in the same ice calorimeter, it will be found to melt exactly the same quantity of ice. This proves these two quantities of heat to be exactly the same, and capable of performing exactly the same amount of physical change. Whatever system of measurement be adopted, their measure must be the same. 138. Measure of Work done in compressing a Gas.--The equivalence between heat and work done, which we have been considering, forms the foundation of the science of heattransfer, or, as it is more commonly called, thermo-dynamics. The possibility of measuring heat by its mechanical equivalent, i.e. by the foot-pound or the erg, reduces the science of heat to a branch of mechanics. To make this clearer, we will consider a particular case, although it can never be fully realized in practice. Suppose a vertical cylinder closed below and also closed above (Fig. 92) by a...

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